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Mg^{2+} ion magnésium

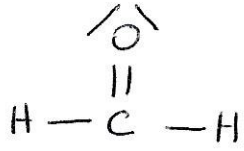
Cl^{-} ion chlore

Na^{+} ion sodium

Ca^{2+} ion calcium

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CH_2O

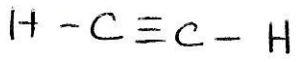


H 1 doublet liant = $2e^{-}$
de valence donc stable

C 4 doublets liants
= $8e^{-}$ de valence donc
stable

O 2 doublets liants
2 doublets non liants
= $8e^{-}$ de valence donc
stable

C_2H_2



H 1 doublet liant = $2e^{-}$
de valence donc stable

C 4 doublets liants
= $8e^{-}$ de valence donc
stable

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1 doublet liant
= $2e^{-}$ de valence
= stable

1 doublet liant
3 doublets non liant
= $8e^{-}$ de valence
= stable

2 doublets liants
2 doublets non liants
= $8e^{-}$ de valence
= stable